**Instructors:** H. S. Biswal and U. Lourderaj **Lab hours:** F M 14:30 hrs to 17:30 hrs

Email: himansu, u.lourderaj

## C343: Physical Chemistry Lab II

## List of proposed experiments

- 1. Synthesis and characterization of flourescein.
- 2. Quantum Yield Calculation for Anthracene.
- 3. An Experiment to determine the Energy of Activation  $(E_a)$
- 4. Determination of viscosity average molecular weight of polyvinyl alcohol, and the fractions of head-to-head monomer linkages in the polymer.
- 5. Static and Dynamic Fluoresence quenching and verification of the Stern-Volmer relationship. Effect of KI on Pyrenesulfonic acid's quantum yield.
- 6. Changes in the absorption and emission spetrum of NBD-octylamine in different solvents.
- 7. Study of the excited state properties of 2-Naphthol : (a) excited state acidity constant;(b) deprotonation and protonation rate constants in the excited state.
- 8. Visualization of atomic orbitals and molecular orbitals.
- 9. Modeling of elimination/migration reaction of Isopropylazide.
- 10. Computation of the transition state for simple reactions.
- 11. Structure and vibrational frequency analysis of hydrogen bonded clusters: Water dimer and water trimer structures, naphthal-water hetero clusters.
- 12. Computation of potential energy profile for the isomerization of n-butane.
- 13. Calculation of thermodynamic parameters ( $\Delta G^o, \Delta H^o, \Delta S^o$ ) for the equilibrium N<sub>2</sub>O<sub>4</sub>  $\rightleftharpoons$  2NO<sub>2</sub>.
- 14. Computational studies on the ground state and excited state properties of some simple molecules.
- 15. Chemical Dynamics Simulations of simple reactions.

## Recommendend Books:

- 1. Experimental Physical Chemistry, R. C. Das and B. Behera, Tata McGraw Hill, 1983.
- 2. A Collection of Interesting General Chemistry Experiments, A. J. Elias, Universities Press, 2007.
- 3. Experimental Physical Chemistry, V. D. Athawale, P. Mathur, New Age International Publishers, 2001.
- 4. Experimental Physical Chemistry: A Laboratory Textbook, A. M. Halpern, Prentice Hall, 2nd edition, 1997.
- 5. G. C. McBane, 3rd edition., W. H. Freeman and Company, New York, 2006.
- 6. http://cdssim.chem.ttu.edu

## **Evaluation/Grading**

You will have to try very hard to fail in this course.

- Quiz: ONE (10%)
- Viva: TWO (20%)
- Continuous evaluation of experiments : 10-12 (40%)
- End-semester exam: ONE (30%)